

Total Synthesis of Shishijimicin A

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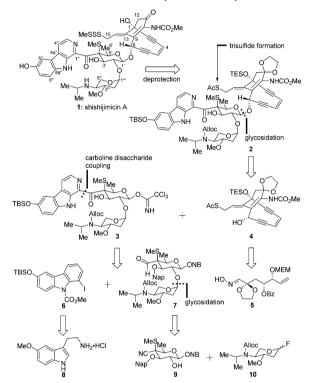
Supporting Information

ABSTRACT: The total synthesis of the rare but extremely potent antitumor agent shishijimicin A has been achieved via a convergent strategy involving carboline disaccharide **3** and hydroxy enediyne thioacetate **4**.

aturally occurring substances provided the first medications to treat disease and continue to be a rich source and inspiration for drug discovery and development.^{1,2} Among them are cytotoxic compounds, some of which are used to treat cancer.¹ Unfortunately, most cytotoxic compounds, including those with the highest potencies, fail to make it to the clinic because of severe side effects. The advent of antibody-drug conjugates (ADCs) ushered in by Mylotarg, the first ADC to be approved by the FDA and equipped with the highly potent natural product calicheamicin $\gamma_1^{1,3,4}$ led to a new paradigm whereby previously failed potent cytotoxic agents became highly desirable as payloads for conjugation to antibodies as potential targeted drugs against cancer. ^{5,6} Shishijimicin A (1) (Scheme 1)⁷ is a rare marine natural product endowed with extremely potent antitumor properties (IC50 = 0.48 pM against P388 leukemia cells).⁷ In view of its phenomenal biological activity and the new paradigm for targeted cancer chemotherapy, shishijimicin A constitutes a highly desirable payload for ADCs, an attractive possibility thus far hindered by its scarcity. In this Communication, we report the first total synthesis of shishijimicin A, rendering the molecule readily available for further biological investigations and opening the way for the construction of designed analogues of this valuable but rare molecule.

The similarities between 1^7 and calicheamicin $\gamma_1^{\ I\ 3,4}$ extend from their common enediyne moiety to their Bergman cycloaromatization-based mechanism of action involving double-stranded DNA cleavage.8 Their structures, however, differ substantially with regard to the constitution of their pentacyclic DNA binding domains, which include a carboline system for shishijimicin A and a fully substituted iodophenyl ring for calicheamicin γ_1^{I} , both of which are known DNA-binding structural motifs.^{9,10} The synthetic roadmap toward 1 was designed on the basis of the retrosynthetic analysis shown in Scheme 1. Thus, protection of the phenolic (TBS), amino (Alloc), and tertiary hydroxyl (TES) groups and transformation of the methyl trisulfide of the target molecule to a thioacetate moiety in the retrosynthetic sense led to its protected enediyne thioacetate 2 as a potential precursor. Disconnection of the glycoside bond linking the enediyne domain of 2 with its pentacyclic appendage revealed enediyne fragment 4 and trichloroacetimidate 3 as potential advanced intermediates for coupling in the synthetic direction. Enediyne 4 was traced back to the readily available key building block 5, which was previously

Scheme 1. Structure and Retrosynthetic Analysis of 1^{a}



^{*a*}Abbreviations: TBS = *tert*-butyldimethylsilyl; TES = triethylsilyl; Ac = acetyl; Alloc = allyloxycarbonyl; Bz = benzoyl; MEM = (2-methoxyethoxy)methyl; Nap =2-naphthylmethyl; NB = *o*-nitrobenzyl.

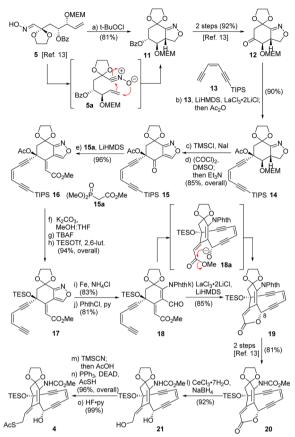
employed in the total synthesis^{11–14} of calicheamicin $\gamma_1^{1.15}$ The pentacyclic advanced intermediate 3 was further disconnected at the indicated carbon–carbon bond bridging the carboline structural motif to the disaccharide domain, furnishing iodocarboline 6 and disaccharide aldehyde 7 [upon modification of the trichloroacetimidate group to the photolabile *o*-nitrobenzyl (NB) ether protecting group] as potential precursors. Finally, 6 was traced back to tryptamine derivative 8, while 7 was disconnected into its obvious monosaccharide units 9 (acceptor) and 10 (donor) as key building blocks.¹⁶

The synthesis of the required enediyne thioacetate precursor 4 from key building block 5^{13} proceeded as shown in Scheme 2. This route represents a streamlined and significantly improved version of the original synthesis of the benzoate counterpart of 4 employed in the total synthesis of calicheamicin γ_1^{11} (19 steps,

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Scheme 2. Synthesis of 4^a



^aReagents and conditions: (a) t-BuOCl (3.0 equiv), benzene, 25 °C, 30 min, 81%; (b) 13 (3.0 equiv), LiHMDS (2.8 equiv), LaCl₃·2LiCl (5.0 equiv), THF, -78 °C, 30 min, then 12, -78 °C, 30 min, then Ac₂O (10.0 equiv), -78 to 25 °C, 2 h, 90%; (c) TMSCl (4.0 equiv), NaI (2.0 equiv), MeCN, 0 to 25 °C, 30 min; (d) (COCl)₂ (4.0 equiv), DMSO (8.0 equiv), CH₂Cl₂, -78 °C, 30 min, then Et₃N (10.0 equiv), -78 to 25 °C, 1.5 h, 85% for the two steps; (e) 15a (2.0 equiv), LiHMDS (1.5 equiv), THF, -78 to 25 °C, 1.5 h, 96%; (f) K₂CO₃ (1.0 equiv), MeOH/THF (1:1), 0 to 25 °C, 3 h; (g) TBAF (1.0 equiv), THF, 0 °C, 10 min; (h) TESOTf (1.5 equiv), 2,6-lutidine (2.0 equiv), CH₂Cl₂, 0 to 25 °C, 2 h, 94% for the three steps; (i) Fe (25 equiv), NH₄Cl (50 equiv), EtOH/H₂O (1:1), 60 °C, 8 h, 83%; (j) PhthCl (1.5 equiv), py (4.0 equiv), MeNO₂, 0 °C, 30 min, 81%; (k) LiHMDS (2.0 equiv), LaCl₃·2LiCl (3.0 equiv), THF, -78 °C, 1 h, 85%; (1) NaBH₄ (2.0 equiv), CeCl₃·7H₂O (3.0 equiv), MeOH, 25 °C, 2 h, 92%; (m) TMSCN (neat), 25 °C, 30 min, then remove excess TMSCN, then dissolve in THF/H₂O (5:1), AcOH (5.0 equiv), 0 °C, 30 min; (n) PPh₃ (5.0 equiv), DEAD (5.0 equiv), AcSH (5.0 equiv), THF, 0 °C, 5 min, 96% for the two steps; (o) HF·py/THF (1:20), 0 °C, 30 min, 99%. LiHMDS = lithium bis(trimethylsilyl)amide; TMS = trimethylsilyl; DMSO = dimethyl sulfoxide; TBAF = tetra-n-butyl ammonium fluoride; Phth = phthaloyl; py = pyridine; DEAD = diethyl azodicarboxylate.

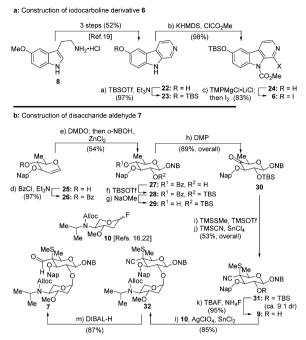
21% overall yield from **5** vs 21 steps, 1.7% overall yield from **5**).¹¹⁻¹⁴ It should also be noted that thioacetate **4** is a more advanced precursor for the methyl trisulfide unit required for both shishijimicin A and calicheamicin γ_1^{I} , thereby saving steps in the postcoupling sequence to the final target. Thus, as shown in Scheme 2, oxidation of oxime **5** to the corresponding nitrile oxide (**5a**) with the improved conditions involving *t*-BuOCl followed by spontaneous [3 + 2] dipolar cycloaddition of the latter intermediate (see **5a** in Scheme 2) led to **11** in 81% overall yield with \geq 10:1 dr (compared with 51% yield and ca. 4:1 dr in the

previous route).¹³ Conversion of the latter compound to ketone 12 (deprotection/oxidation) proceeded smoothly as previously reported (two steps, 92% overall yield).¹³ The subsequent coupling of 12 with enediyne fragment 13,¹⁷ however, was significantly improved by using LiHMDS in the presence of LaCl₃·2LiCl,¹⁸ affording, after in situ acetylation, the desired enedivne 14 in 90% overall yield (compared with 69% yield under the originally employed conditions).¹³ Removal of the MEM group from 14 followed by Swern oxidation of the resulting secondary alcohol and concomitant oxidation of the isoxazoline to the isoxazole moiety furnished keto-isoxazole 15 (85% overall yield for the two steps). The latter intermediate served admirably as a substrate for the exclusively E-selective Horner-Wadsworth-Emmons olefination that followed [(MeO)₂P(O)CH₂CO₂Me (15a), LiHMDS], leading to $E-\alpha_{\beta}\beta$ unsaturated methyl ester 16 in 96% yield. Acetate 16 was then transformed to terminal acetylene TES ether 17 in 94% overall yield through a sequence involving removal of the acetate group (K₂CO₃), cleavage of the TIPS moiety (TBAF), and silvlation (TESOTf). Rupture of the isoxazole moiety in 17 was then achieved more conveniently and efficiently than before¹³ through the use of Fe in EtOH/ H_2O (83%), and the resulting amino aldehyde was captured by phthaloyl chloride (PhthCl) in the presence of pyridine to afford N-phthalide aldehyde 18 (81%). The direct and stereoselective cyclization of 18 to give cyclic enediyne 19 (via intermediate 18a; see Scheme 2) using LiHMDS-LaCl₃·2LiCl¹⁸ in THF in 85% yield represents a major improvement over the previously used three-step sequence requiring inversion of the opposite configuration at C8 obtained from the same substrate (i.e., 18) through the use of KHMDS in toluene.¹³ The observed stereoselectivity is presumed to be due to complexation of La³⁺ to the aldehyde and ester oxygens, which fixes the conformation of the aldehyde moiety in the proper orientation. The N-phthalide moiety of 19 was then converted to the desired methyl carbamate group by reaction with MeNHNH₂ followed by exposure of the resulting amine to triphosgene in the presence of pyridine and MeOH as previously reported,¹³ affording enediyne lactone 20 in 81% overall yield. Reduction of the lactone moiety in 20 was achieved in one step and 92% yield using NaBH₄-CeCl₃·7H₂O (as opposed to two steps and 84% overall yield in the original route),¹³ providing a further improvement in the overall sequence to enediyne diol 21. Finally, conversion of 21 to the targeted enediyne thioacetate fragment 4 was accomplished efficiently by sequential treatment with excess TMSCN (bissilvlation), AcOH (selective primary TMS cleavage), Ph₃P-DEAD-AcSH (Mitsunobu reaction, thioacetate formation), and HF·py (secondary TMS cleavage) in 95% overall yield.

To construct iodocarboline 6 (Scheme 3a), carboline 22 [prepared in 52% overall yield from commercially available 5methoxytryptamine (8) through a known three-step sequence¹⁹] was silylated (TBSOTf, Et₃N, 97% yield) to afford 23, which was converted to carbamate 24 (KHMDS, ClCO₂Me, 98% yield). The latter compound was reacted with 2,2,6,6-tetramethylpiperidinylmagnesium chloride·lithium chloride complex (TMPMgCl·LiCl)²⁰ and I₂, furnishing the desired iodocarboline 6 in 83% yield.

The required disaccharide 7 was synthesized from the readily available glucal 25^{21} and glycosyl fluoride $10^{16,22}$ as depicted in Scheme 3b. Benzoylation of the free hydroxyl group of 25 (BzCl, Et₃N, 97% yield) followed by sequential treatment of the resulting benzoate glucal 26 with in situ-generated DMDO and *o*-nitrobenzyl alcohol (*o*-NBOH) furnished hydroxy-*o*-nitro-

Scheme 3. Syntheses of 6 and 7^a

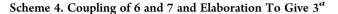


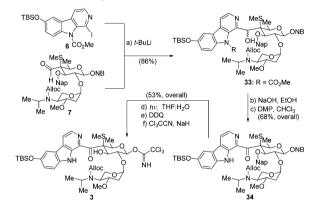
^aReagents and conditions: (a) TBSOTf (1.05 equiv), Et₃N (3.0 equiv), DMF, 0 °C, 30 min, 97%; (b) ClCO₂Me (1.1 equiv), KHMDS (1.05 equiv), THF, 0 °C, 30 min, 98%; (c) TMPMgCl·LiCl (4.0 equiv), THF, -78 to 25 °C, 4 h, then I₂ (5.0 equiv), THF, -78 to 0 °C, 30 min, 83%; (d) BzCl (1.05 equiv), Et₃N (1.1 equiv), CH₂Cl₂, 0 °C, 30 min, 97%; (e) Oxone (5.0 equiv), NaHCO₃ (25 equiv), acetone/H2O/CH2Cl2 (1:3:4), 25 °C, 4 h, then o-NBOH (3.0 equiv), ZnCl₂ (1.5 equiv), 4 Å MS, THF, -78 to 25 °C, 4 h, 54%; (f) TBSOTf (1.05 equiv), Et₃N (1.1 equiv), CH₂Cl₂, 0 °C, 30 min; (g) NaOMe (10.0 equiv), MeOH, 40 °C, 24 h; (h) DMP (1.2 equiv), CH₂Cl₂, 0 to 25 °C, 1 h, 89% for the three steps; (i) TMSSMe (2.5 equiv), TMSOTf (1.5 equiv), toluene, -20 to 0 °C, 30 min, 61%; (j) TMSCN (3.5 equiv), SnCl₄ (1.5 equiv), CH₂Cl₂, 0 °C, 3 h, 87% (ca. 9:1 dr); (k) TBAF (5.0 equiv), NH₄F (10.0 equiv), THF, 0 °C, 1 h, 95%; (1) 10 (2.2 equiv), AgClO₄ (2.5 equiv), SnCl₂ (2.5 equiv), 4 Å MS, THF, -78 to 25 °C, 12 h, 85%; (m) DIBAL-H (3.0 equiv), CH_2Cl_2 , -78 °C, 45 min, 87%. KHMDS = potassium bis-(trimethylsilyl)amide; TMP = 2,2,6,6-tetramethylpiperidinyl; DMDO = dimethyldioxirane; o-NBOH = o-nitrobenzyl alcohol; DMP = Dess-Martin periodinane; DIBAL-H = diisobutylaluminum hydride.

benzyl ether 27 in 54% overall yield via the corresponding epoxide intermediate.²³ The newly generated hydroxyl group of 27 was converted to its TBS ether (TBSOTf, Et₃N) to afford 28, from which the benzoate moiety was cleaved (NaOMe) to give alcohol 29. DMP oxidation of 29 led to ketone 30 in 89% overall yield for the three steps from 27. Reaction of 30 with TMSSMe in the presence of TMSOTf furnished the corresponding methylthioketal,^{16,24} which underwent stereoselective cyanation²⁵ upon exposure to TMSCN and SnCl₄ to afford nitriles 31 and 4-epi-31 (ca. 9:1 dr, 53% yield for the two steps). Removal of the TBS group from 31 (TBAF, NH_4F)²⁶ gave carbohydrate acceptor 9 (95% yield), whose coupling with carbohydrate donor $10^{16,22}$ proceeded smoothly in the presence of AgClO₄ and SnCl₂ to afford the desired α -glycoside 32 stereoselectively in 85% yield. Finally, DIBAL-H reduction of the nitrile group in 32 led to the targeted aldehyde 7 in 87% yield.

The coupling of iodocarboline **6** and disaccharide aldehyde 7 proceeded through the lithio derivative of the former (generated with *t*-BuLi at -78 °C) to give alcohol **33** (86% yield, ca. 1:1 dr,

inconsequential), as shown in Scheme 4. Treatment of carboline carbamate 33 with NaOH in EtOH led to the corresponding free amine, whose oxidation with DMP gave ketone 34 in 68% yield over the two steps. Photolytic cleavage¹⁴ of the *o*-nitrobenzyl ether moiety from the latter compound followed by sequential treatment with DDQ (removal of naphthyl group) and Cl_3CCN –NaH (trichloroacetimidate formation) resulted in the stereoselective formation of the coveted trichloroacetimidate 3 (53% overall yield, β -anomer exclusively).



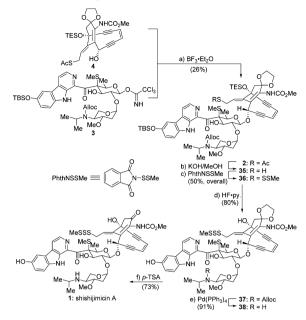


^aReagents and conditions: (a) **6** (3.0 equiv), *t*-BuLi (6.0 equiv), THF, -78 °C, 30 min, then 7 (1.0 equiv), -78 to -35 °C, 40 min, 86% (ca. 1:1 dr) based on 7; (b) NaOH (3.0 equiv), EtOH, 0 to 25 °C, 2.5 h; (c) DMP (1.1 equiv), CHCl₃, 0 to 35 °C, 10 min, 68% for the two steps; (d) $h\nu$, THF/H₂O (10:1), 4.5 h; (e) DDQ (2.5 equiv), CH₂Cl₂/H₂O (10:1), 30 °C, 1.5 h; (f) NaH (2.0 equiv), Cl₃CCN/ CH₂Cl₂ (1:2), 25 °C, 5 min, 53% for the three steps. DDQ = 2,3dichloro-5,6-dicyano-1,4-benzoquinone.

Having assembled the two advanced intermediates, trichloroacetimidate 3 and hydroxy enediyne 4, the next objective became their coupling and elaboration of the resulting product to give shishijimicin A (1). Scheme 5 depicts how this challenging task was accomplished. Indeed, it was after considerable experimentation that the two fragments (i.e., 3 and 4) were joined through the action of BF₃·Et₂O to afford β -glycoside 2 selectively in 26% vield. It should be noted that the corresponding naphthyl ether trichloroacetimidate proved resistant to glycosidation, presumably because of severe steric hindrance, an effect also assumed to be responsible for the rather low yield observed for the reaction between 3 and 4. Improvement of this coupling reaction ought to be possible, and studies toward this goal are currently in progress. Enediyne thioacetate 2 was transformed to the protected form of shishijimicin A, precursor 36, through sequential treatment with KOH in MeOH (acetate cleavage) and N-methyldithiophthalimide (PhthNSSMe)²⁷ in 50% overall yield via thiol derivative 35. Desilylation of 36 with HF·py furnished advanced intermediate 37 (80% yield), from which the Alloc protecting group was removed by exposure to catalytic $Pd(PPh_3)_4$, leading to the shishijimicin A penultimate precursor ketal 38 (91% yield). Finally, cleavage of the ketal moiety from precursor 38 gave the targeted natural product, shishijimicin A (1), in 73% yield. The physical data for synthetic 1 matched those reported for the natural substance.

In conclusion, we have developed a convergent and modular strategy for the total synthesis of the rare and precious marine natural product shishijimicin A (1). The reported synthesis

Scheme 5. Coupling of 3 and 4 and Total Synthesis of 1^a



^aReagents and conditions: (a) 3 (1.0 equiv), 4 (1.6 equiv), BF₃·OEt₂ (3.5 equiv), 4 Å MS, CH₂Cl₂, -78 to -40 °C, 1 h, 26% based on 3; (b) KOH (10.0 equiv), MeOH, -5 °C, 1.5 h, then AcOH (10.0 equiv); (c) PhthNSSMe (6.0 equiv), CH₂Cl₂, 0 °C, 15 min, 50% for the two steps; (d) HF·py/THF (1:20), 0 to 25 °C, 4 h, 80%; (e) Pd(PPh₃)₄ (0.5 equiv), morpholine (15 equiv), THF, 0 °C, 45 min, 91%; (f) *p*-TSA (3.0 equiv), THF/acetone/H₂O (20:20:1), 25 °C, 48 h, 73%. *p*-TSA = *p*-toluenesulfonic acid.

makes this potent antitumor agent available for further biological investigations and potential medical applications, including the design and synthesis of ADCs and other conjugate assemblies for which shishijimicin A may serve as a uniquely powerful payload. Improvements and applications of the described chemistry could now be envisioned for the synthesis of congeners^{7,28} and analogues of **1** as potentially useful molecular entities for incorporation into these types of targeted and personalized cancer chemotherapeutic agents.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures and compound characterization. The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.5b05575.

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Notes

The authors declare no competing financial interest.

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